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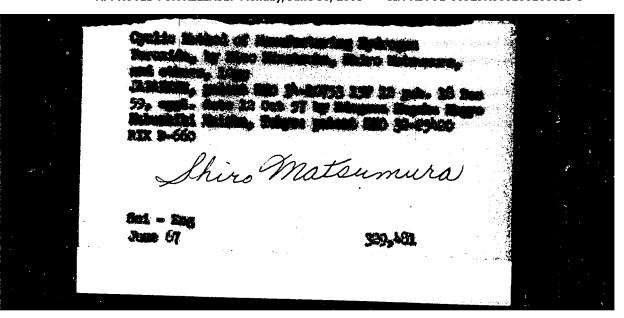
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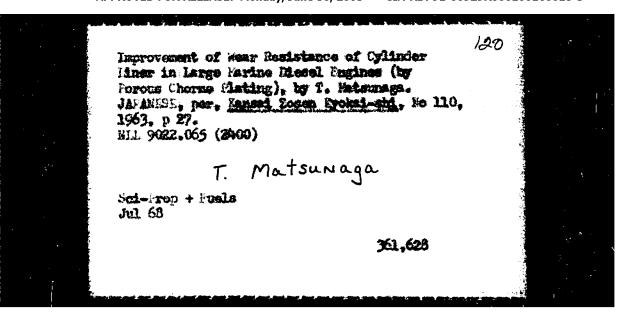
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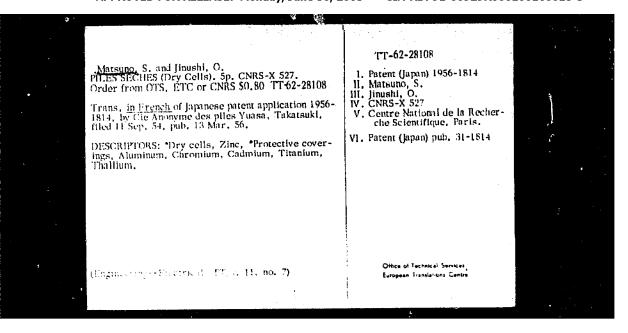
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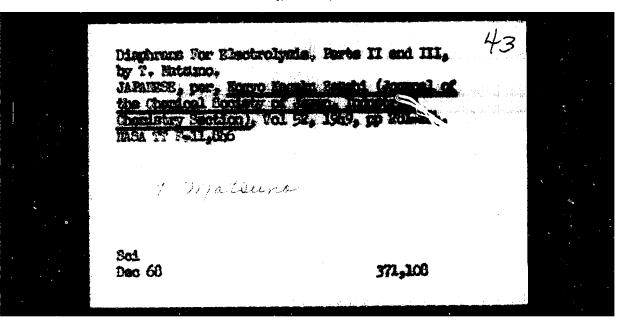
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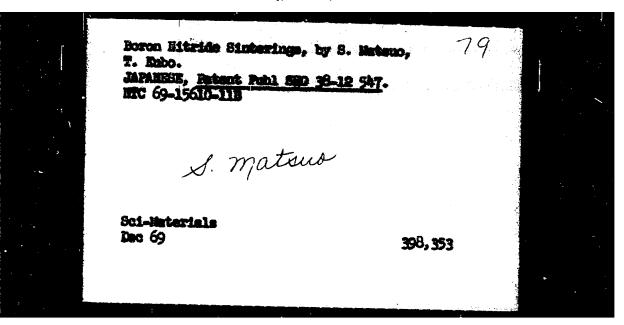
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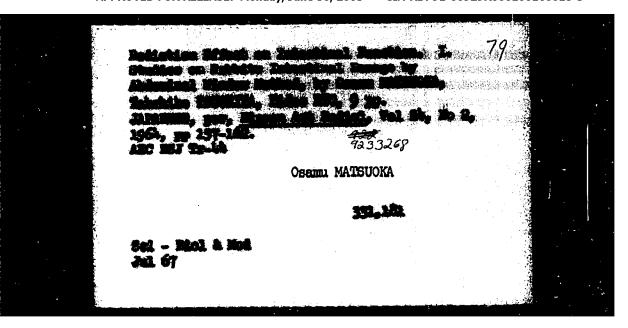
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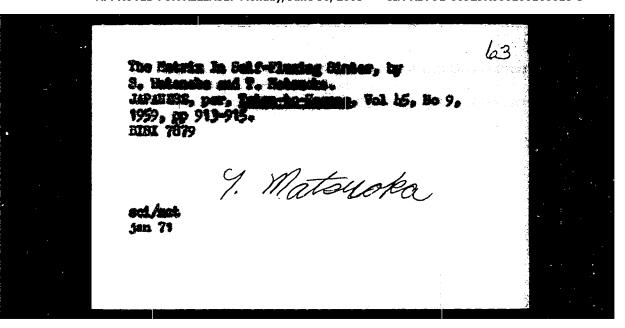
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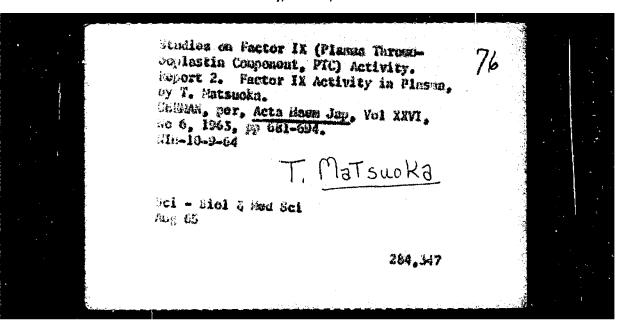
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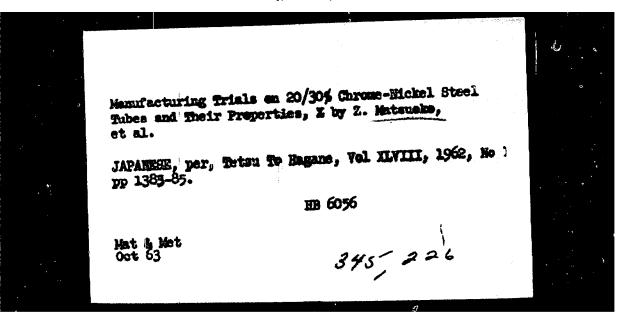
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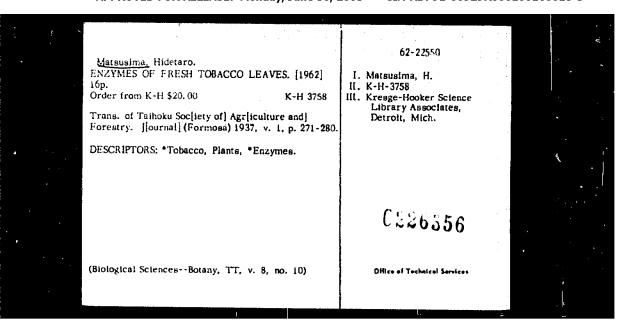
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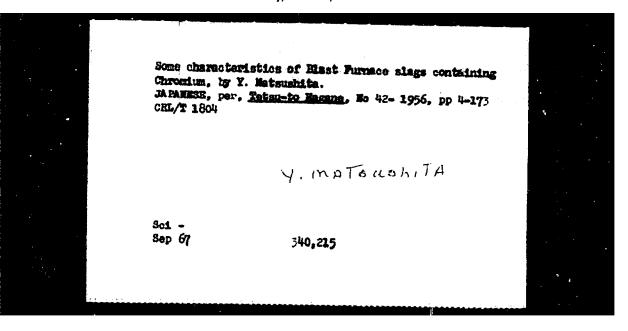
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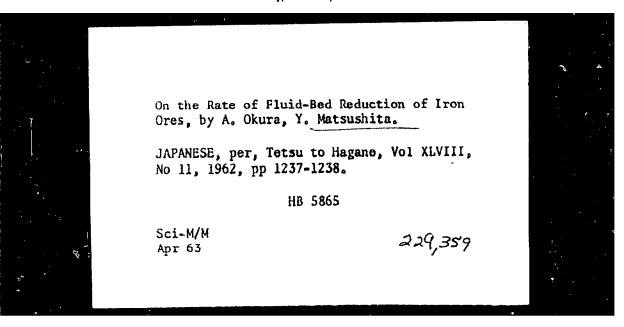


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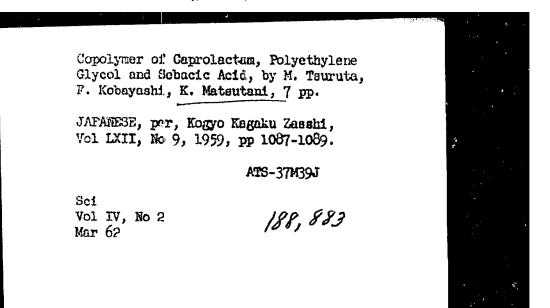


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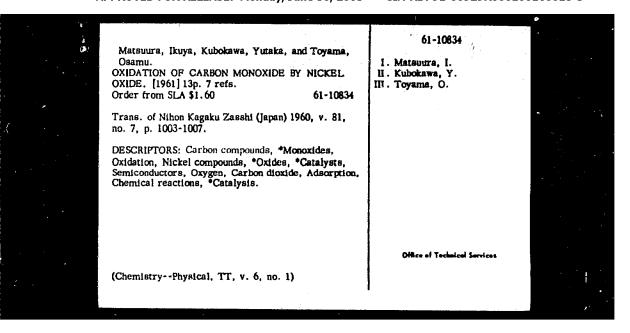
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furnaces, *Zinc compounds, Oxides, Separation,
Pelleta, Coke, Rotary furnaces.

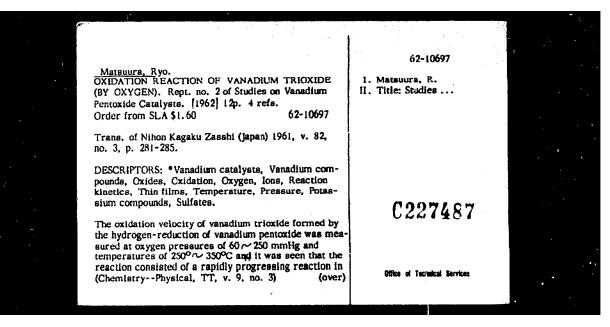
In this method of removing Zn from open hearth furnace
dust, a product of semi-reduced Luppen of about 60%
total iron (40% metallic Fe), Zn < 3.0% and its byproducts (about 35% of zinc oxide) were obtained. This
zinc oxide may be effectively utilized and therefore, it
is well justified budget-wise. An important reason for
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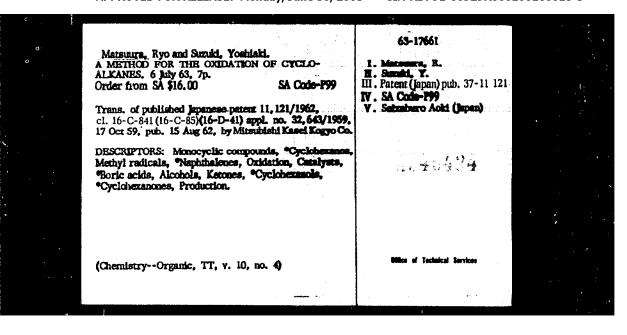
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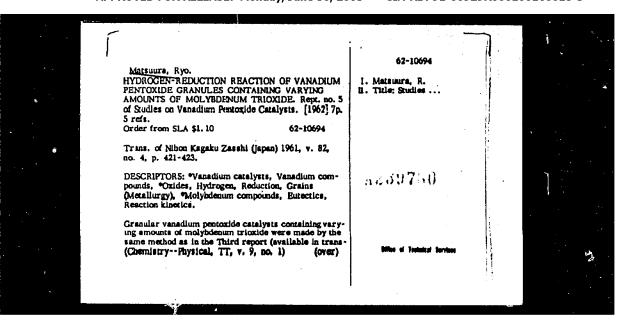
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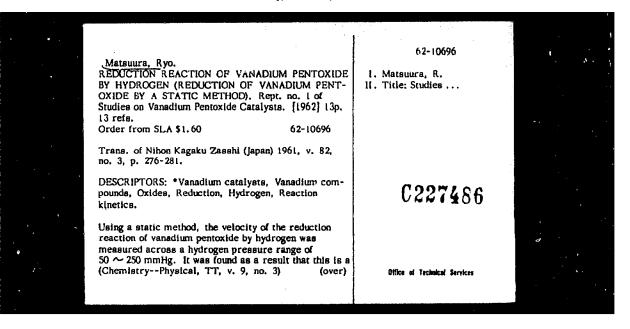
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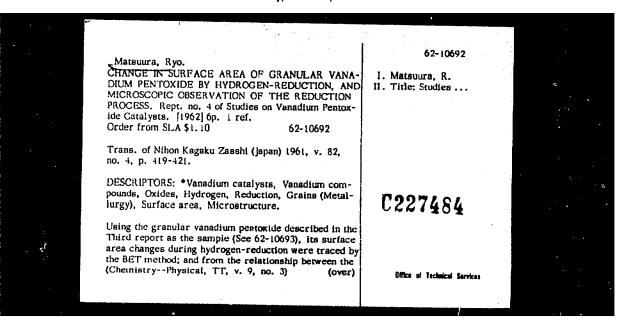
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Spherical vanadium pentoxide samples were prepared, and these were reduced by the normal-pressure flow-through method; the reduction yelocity was determined from the amount of water formed. It was discovered as (Chemistry--Physical, TT, v. 9, no. 3) (over)

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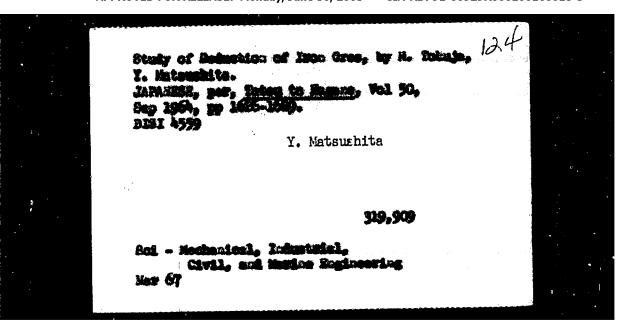
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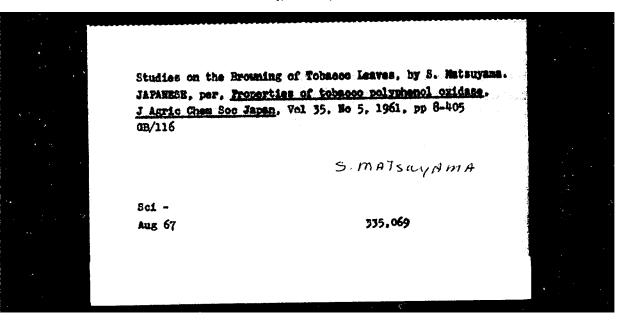
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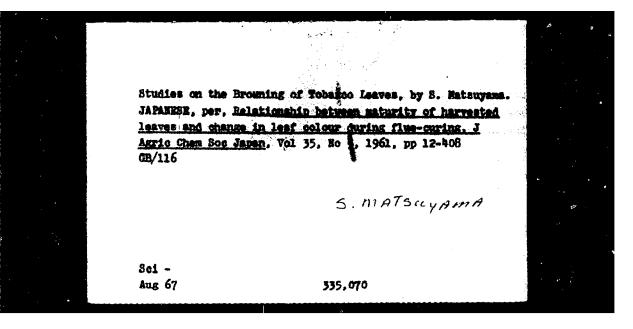
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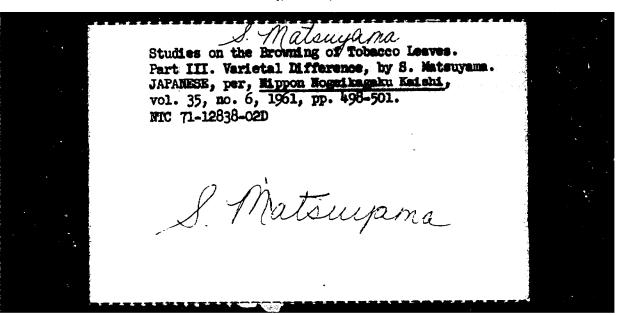


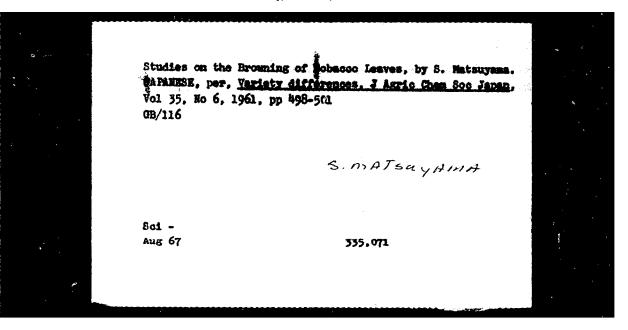
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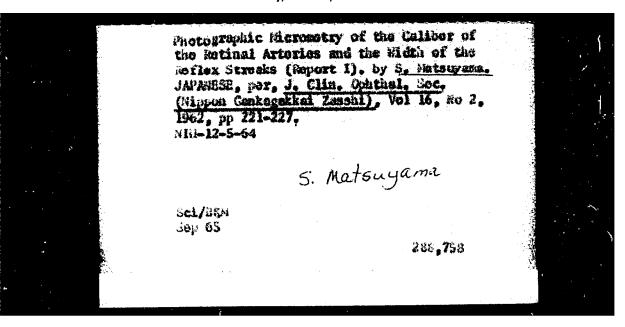
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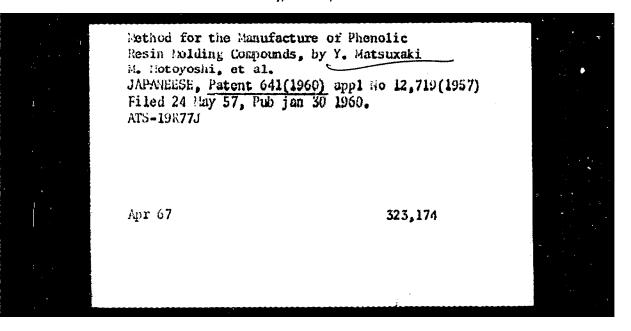
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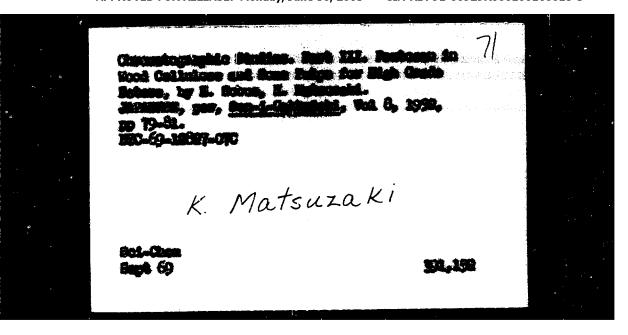
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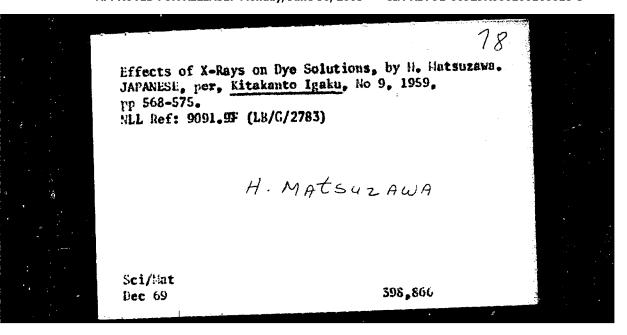
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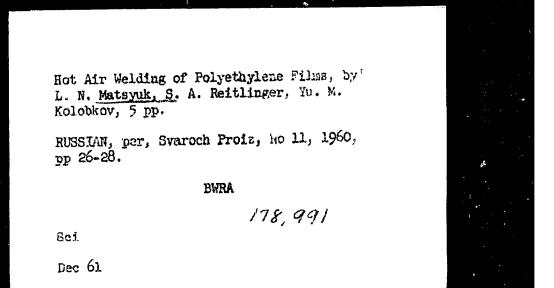
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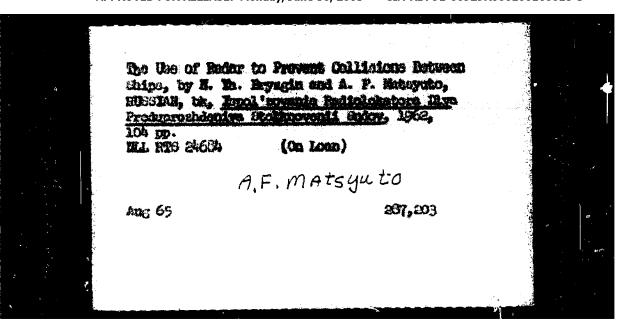
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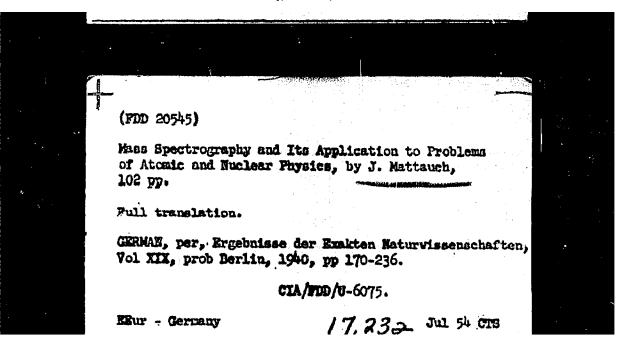
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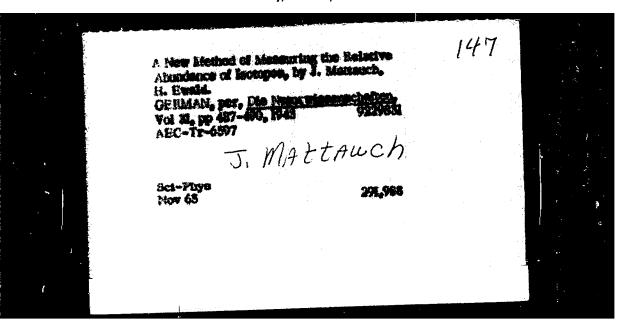
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